

## Chapter 6 Solutions Thermodynamics An Engineering Approach 7th

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**Thermodynamics Q6.7 Chapter 6 Class 11 CHEMISTRY NCERT Solutions** In a process, 701 J of heat is absorbed by a system and 394 J of work is done by the system. What is the change in internal energy ...

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**Thermodynamics Q6.18 Chapter 6 Class 11 CHEMISTRY NCERT Solutions** For the reaction,  $2 \text{Cl} (\text{g}) \rightarrow \text{Cl}_2 (\text{g})$ , what are the signs of  $\Delta H$  and  $\Delta S$ ? Visit our facebook page ...

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**Thermodynamics Q6.12 Chapter 6 Class 11 CHEMISTRY NCERT Solutions** Enthalpies of formation of  $\text{CO}(\text{g})$ ,  $\text{CO}_2(\text{g})$ ,  $\text{N}_2\text{O}(\text{g})$  and  $\text{N}_2\text{O}_4(\text{g})$  are -110, -393, 81 and 9.7 kJ mol<sup>-1</sup> respectively. Find the value ...

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**Thermodynamics Q6.5 Chapter 6 Class 11 CHEMISTRY NCERT Solutions** The enthalpy of combustion of methane, graphite and dihydrogen at 298 K are, -890.3 kJ mol<sup>-1</sup>, -393.5 kJ mol<sup>-1</sup>, and -285.8 kJ ...

**Thermodynamics Q6.21 Chapter 6 Class 11 CHEMISTRY NCERT Solutions** Comment on the **thermodynamic** stability of  $\text{NO}(\text{g})$ , given  $\frac{1}{2} \text{N}_2 + \frac{1}{2} \text{O}_2(\text{g}) \rightarrow \text{NO}(\text{g})$ ;  $\Delta_r H^\circ = 90 \text{ kJ mol}^{-1}$   $\text{NO} (\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \dots$

**Thermodynamics Q6.8 Chapter 6 Class 11 CHEMISTRY NCERT Solutions** The reaction of cyanamide,  $\text{NH}_2\text{CN} (\text{s})$ , with dioxygen was carried out in a bomb calorimeter, and  $\Delta U$  was found to be -742.7 kJ ...

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